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# - R. Original Contribution

# PHOSPHINE-INDUCED OXIDATIVE DAMAGE IN RATS: ATTENUATION BY MELATONIN

CHING-HUNG HSU,\* BOR-CHENG HAN,\* MING-YIE LIU,<sup>†</sup> CHING-YING YEH,\* and JOHN E. CASIDA<sup>‡</sup>

\*Department of Public Health, School of Medicine, Taipei Medical College, Taipei, Taiwan, ROC; <sup>†</sup>Department of Environmental and Occupational Health, National Cheng Kung University, Medical College, Tainan, Taiwan, ROC; and <sup>‡</sup>Department of Environmental Science, Policy and Management, University of California, Berkeley, CA, USA

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Abstract—Phosphine (PH<sub>3</sub>), from hydrolysis of aluminum, magnesium and zinc phosphide, is an insecticide and rodenticide. Earlier observations on PH<sub>3</sub>-poisoned insects, mammals and a mammalian cell line led to the proposed involvement of oxidative damage in the toxic mechanism. This investigation focused on PH<sub>3</sub>-induced oxidative damage in rats and antioxidants as candidate protective agents. Male Wistar rats were treated ip with PH<sub>3</sub> at 2 mg/kg. Thirty min later the brain, liver, and lung were analyzed for glutathione (GSH) levels and lipid peroxidation (as malondialdehyde and 4-hydroxyalkenals) and brain and lung for 8-hydroxydeoxyguanosine (8-OH-dGuo) in DNA. PH<sub>3</sub> caused a significant decrease in GSH concentration and elevation in lipid peroxidation in brain (36–42%), lung (32–38%) and liver (19–25%) and significant increase of 8-OH-dGuo in DNA of brain (70%) and liver (39%). Antioxidants administered ip 30 min before PH<sub>3</sub> were melatonin, vitamin C, and  $\beta$ -carotene at 10, 30, and 6 mg/kg, respectively. The PH<sub>3</sub>-induced changes were significantly or completely blocked by melatonin while vitamin C and  $\beta$ -carotene were less effective or inactive. These findings establish that PH<sub>3</sub> induces and melatonin protects against oxidative damage in the brain, lung and liver of rats and suggest the involvement of reactive oxygen species in the genotoxicity of PH<sub>3</sub>. © 2000 Elsevier Science Inc.

Keywords—Melatonin, Antioxidants, Pesticide, Phosphine, Free radicals, Glutathione, Lipid peroxidation, 8-Hydroxydeoxyguanosine

# INTRODUCTION

Phosphine (PH<sub>3</sub>) is a widely-used fumigant for the control of stored product insects. It is normally generated by the action of ambient water vapor on a solid formulation containing aluminium or magnesium phosphide mixed with other ingredients designed to regulate the release of the gas. At present the major fumigants for controlling insect pests in stored products are PH<sub>3</sub> and methyl bromide. Because methyl bromide is being phased out because of adverse environmental effects, the role of PH<sub>3</sub> is becoming of even greater importance [1]. Additionally, zinc phosphide is a major rodenticide, liberating PH<sub>3</sub> when ingested [2]. PH<sub>3</sub> is also used in the synthesis of organophosphines and as a dopant in semiconductor production [3].

PH<sub>3</sub> is highly toxic to many animals [3]. AlP, with a rat oral  $LD_{50}$  of 14 mg/kg [4], is responsible for many human poisonings in India [5,6]. Fumigators exposed to PH<sub>3</sub> may have an increased frequency of chromosomal abberations in their peripheral blood lymphocytes [7–9]. PH<sub>3</sub> is weakly genotoxic in mice, i.e., exposure of Balb-c mice (4.5 ppm, 13 weeks) results in significant increases in micronucleus frequency in bone marrow and spleen lymphocytes [10].

 $PH_3$  is a respiratory inhibitor and induces oxidative damage in animals [11]. It inhibits the activities of cytochrome c oxidase [12–14], catalase [15–17] and peroxidase [15,16], stimulates the production of hydrogen peroxide and reactive oxygen species (ROS) [14,18] and elevates superoxide dismutase (SOD) [15,16]. The malondialdehyde (MDA) level is elevated in cardiac tissue of AlP-poisoned rats [19]. Consistent with these obser-

Address correspondence to: Dr. John E. Casida, Environmental Chemistry and Toxicology Laboratory, Department of Environmental Science, Policy and Management, 114 Wellman Hall, University of California, Berkeley, CA 94720-3112, USA; Tel: (510) 642-5424; Fax: (510) 642-6497; E-Mail: ectl@nature.berkeley.edu.

vations, AIP-poisoned humans show significantly higher SOD and MDA levels and lower catalase levels in serum compared with unexposed patients [5]. Intravenous magnesium reportedly reduces oxidative stress and mortality in humans with acute AIP poisoning [6].

Melatonin, a major secretory product of the pineal gland, scavenges hydroxyl radical [20], peroxynitrite [21], singlet oxygen [22], and possibly peroxyl radical [23], which are generated during the oxidation of unsaturated lipids and lead to the propagation of lipid peroxidation. The effectiveness of melatonin is facilitated by its combined lipophilic [24] and hydrophilic [25] character, allowing transport across the blood–brain-barrier and distribution throughout the cell [26]. Melatonin may reduce oxidative stress also by stimulating some important antioxidative enzymes, i.e., SOD [27], glutathione (GSH) reductase [28], glucose-6-phosphate dehydrogenase [29], and GSH peroxidase [30].

PH<sub>3</sub> causes oxidative toxicity in insects, mammals, and mammalian cell lines. However, there are no reports on the effect of antioxidants on PH<sub>3</sub>-induced oxidative stress in mammals. The first aim of the present study is to examine possible PH<sub>3</sub>-induced oxidative damage in rats using as criteria the levels of GSH, GSH disulfide (GSSG), lipid peroxidation products (MDA plus 4-hydroxyalkenals (4-HDA)), and 8-hydroxydeoxyguanosine (8-OH-dGuo) in DNA. The second goal is to compare melatonin with vitamin C and  $\beta$ -carotene as candidate antioxidants to protect against oxidative damage induced by PH<sub>3</sub>.

#### MATERIALS AND METHODS

#### Chemicals and treatment solutions

AlP (> 95% purity) was from Chem Service (West Chester, PA, USA) and other compounds from Sigma Chemical (St. Louis, MO, USA). A saturated aqueous solution of PH<sub>3</sub> was prepared by introducing freshlypowdered AlP (20 mg) into a 30 ml serum vial that was then stoppered with a rubber septum, and saline (20 ml) was injected in through the septum with immediate mixing. Insoluble salts were allowed to precipitate for 2 min before taking up the liquid portion (with some suspended solid) into a hypodermic syringe. This preparation is nominally 1 mg AlP/ml or 0.59 mg PH<sub>3</sub> equivalents/ml. However, some of the PH<sub>3</sub> would be in the gas phase and the solubility of  $PH_3$  in water is 11.6 mM [11] or 0.39 mg PH<sub>3</sub>/ml, which was considered to be the actual concentration. The PH<sub>3</sub> solution was administered ip to rats at 0.5 ml/100 g, equivalent to 2 mg PH<sub>3</sub>/kg body weight. Melatonin was dissolved in ethanol then saline added to 3% final ethanol concentration. Vitamin C and  $\beta$ -carotene were dissolved in saline and corn oil, respectively.

Table 1. Three Sets of Experiments With Different Antioxidants (1–3), Each One Involving 24 Male Wistar Rats, Fasted Overnight, and Randomly Divided Into Four Groups of Six Rats (a–d)

Set	Antioxidant	Group	Carrier solution	Saline
1 2 3	melatonin vitamin C β-carotene	a b c d	alone alone antioxidant antioxidant	alone PH <sub>3</sub> alone PH <sub>3</sub>

The antioxidant solutions were administered ip at 0.5 ml/100 g body weight to yield dosages for melatonin, vitamin C, and  $\beta$ -carotene of 10, 30, and 6 mg/kg, respectively.

### Experimental design (animal treatment)

Male Wistar rats  $(250 \pm 30 \text{ g})$  were maintained on rat chow basal diet and water ad libitum. Three sets of experiments were made with different antioxidants (1–3), each one involving 24 rats fasted overnight and randomly divided into four groups of six rats (a–d) (Table 1).

Each rat received carrier solvent (saline, 3% ethanol, or corn oil) alone or with an antioxidant (vitamin C, melatonin, or  $\beta$ -carotene, respectively) followed after 30 min by saline alone or with PH<sub>3</sub>. The findings with group a were the same in each of the three sets and in group b were also the same in each set, i.e., there was no significant difference in the controls for the three carrier solvents and in the effect of PH<sub>3</sub> following each of the carrier solvent pretreatments.

Rats killed by carbon dioxide anesthesia were subjected to intracardiac perfusion with ice-cold saline to eliminate excess iron that could otherwise be released from intracellular storage sites resulting in an artificial increase in free radical formation [31]. Brain, lung, and liver were removed and immediately frozen on solid carbon dioxide. All tissues were kept at  $-80^{\circ}$ C until the time of the analysis.

## Measurement of GSH and GSSG

A portion of each tissue was homogenized at 20% (w/v) in 1% picric acid in water using a glass-Teflon Potter-Elvehjem tissue grinder. The supernatant from centrifugation (15,000  $\times$  g, 15 min, 4°C) was used to measure GSH and GSSG by a procedure involving GSH reductase, 2-vinylpyridine and 5,5'-dithio-bis(2-nitrobenzoic acid) [32,33].

### Measurement of aldehydic lipid peroxidation products

Tissue was homogenized as above at 10% (w/v) in ice-cold 20 mM Tris-HCl buffer, pH 7.4. The superna-

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Tissue	Antioxidant pretreatment				
Toxicant	None	Melatonin	Vitamin C	β-carotene	
GSH level (ar	nd GSSG level in paren	thesis), $\mu$ mol/g tissue			
Brain	ľ	// B			
Control	$1.45 \pm 0.11$	$1.53 \pm 0.11$	$1.48 \pm 0.07$	$1.52 \pm 0.13$	
	$(0.018 \pm 0.001)$	$(0.017 \pm 0.002)$	$(0.018 \pm 0.001)$	$(0.018 \pm 0.001)$	
PH <sub>3</sub>	$0.93 \pm 0.11^{a}$	$1.53 \pm 0.09^{b}$	$1.13 \pm 0.10^{a}$	$0.98 \pm 0.10^{a}$	
5	$(0.026 \pm 0.002^{\rm a})$	$(0.017 \pm 0.001^{\rm b})$	$(0.022 \pm 0.001^{\rm a})$	$(0.023 \pm 0.002^{\rm a})$	
Lung	· /			· · · · · ·	
Control	$1.30 \pm 0.20$	$1.33 \pm 0.16$	$1.30 \pm 0.10$	$1.33 \pm 0.11$	
	$(0.038 \pm 0.002)$	$(0.036 \pm 0.002)$	$(0.037 \pm 0.002)$	$(0.037 \pm 0.001)$	
$PH_3$	$0.80 \pm 0.09^{a}$	$1.20 \pm 0.07^{b}$	$1.08 \pm 0.09$	$1.00 \pm 0.10$	
5	$(0.051 \pm 0.003^{a})$	$(0.036 \pm 0.002^{\rm b})$	$(0.043 \pm 0.002)$	$(0.045 \pm 0.003^{a})$	
Liver					
Control	$4.40 \pm 0.18$	$4.38 \pm 0.22$	$4.42 \pm 0.11$	$4.38 \pm 0.12$	
	$(0.110 \pm 0.004)$	$(0.108 \pm 0.003)$	$(0.112 \pm 0.003)$	$(0.109 \pm 0.005)$	
$PH_3$	$3.58 \pm 0.20^{\rm a}$	$4.40 \pm 0.09^{b}$	$4.00 \pm 0.18$	$3.93 \pm 0.12^{a}$	
5	$(0.130 \pm 0.003^{a})$	$(0.114 \pm 0.006^{\rm b})$	$(0.119 \pm 0.004)$	$(0.122 \pm 0.004)$	
MDA + 4-HI	DA level, nmol/mg prot	ein			
Brain					
Control	$3.00 \pm 0.07$	$2.80 \pm 0.09$	$2.99 \pm 0.10$	$3.03 \pm 0.11$	
$PH_3$	$4.27 \pm 0.10^{a}$	$3.10 \pm 0.09^{b}$	$4.01 \pm 0.20^{a}$	$4.12 \pm 0.13^{a}$	
Lung					
Control	$2.58 \pm 0.09$	$2.57 \pm 0.17$	$2.58 \pm 0.14$	$2.68 \pm 0.14$	
PH <sub>3</sub>	$3.40 \pm 0.20^{a}$	$2.58 \pm 0.09^{b}$	$3.01 \pm 0.18$	$3.21 \pm 0.17^{a}$	
Liver					
Control	$0.81 \pm 0.02$	$0.78 \pm 0.06$	$0.80 \pm 0.03$	$0.90 \pm 0.04$	
PH <sub>3</sub>	$1.01 \pm 0.05^{a}$	$0.80 \pm 0.03^{\text{b}}$	$0.80 \pm 0.10$	$0.89 \pm 0.14$	
8-OH-dGuo/1	0 <sup>5</sup> dGuo ratio				
Brain					
Control	$3.18 \pm 0.25$	$2.72 \pm 0.21$	$3.20 \pm 0.28$	$3.10 \pm 0.36$	
PH <sub>3</sub>	$5.40 \pm 0.47^{a}$	$3.02 \pm 0.37^{6}$	$4.48 \pm 0.40^{a}$	$5.02 \pm 0.32^{a}$	
Liver					
Control	$4.30 \pm 0.37$	$3.53 \pm 0.39$	$4.25 \pm 0.40$	$4.12 \pm 0.44$	
PH <sub>3</sub>	$5.98 \pm 0.63^{a}$	$4.00 \pm 0.32^{6}$	$4.67 \pm 0.50$	$5.50 \pm 0.46$	

Table 2. Effects of PH<sub>3</sub> and Antioxidants on Levels of GSH, GSSG, Lipid Peroxidation Products and 8-OH-dGuo/dGuo Ratio in IP-Treated Rats

Rats were treated ip with PH<sub>3</sub> at 2 mg/kg and tissues analyzed 30 min later. Antioxidant pretreatment (melatonin, vitamin C, or  $\beta$ -carotene at 10, 30, or 6 mg/kg, respectively) was 30 min before PH<sub>3</sub>. For conditions see Materials and Methods. The tabulated results in the left column (none) are for pretreatments with saline but the findings with 3% ethanol or corn oil are not significantly different and in fact are practically identical (data not given). Data are mean  $\pm$  SEM, n = 6.

<sup>a</sup>  $PH_3$  significantly different from corresponding control without  $PH_3$  (p < .05).

<sup>b</sup> Melatonin with PH<sub>3</sub> significantly different from no antioxidant with PH<sub>3</sub> (p < .05).

tant from centrifugation  $(2500 \times g, 30 \text{ min}, 4^{\circ}\text{C})$  was analyzed for MDA plus 4-HDA using the lipid peroxidation kit of Calbiochem-Novabiochem Corp. (La Jolla, CA, USA) [18] as an estimate of lipid peroxidation [34]. The protein concentration was determined [35] with bovine serum albumin as the standard.

### Measurement of 8-OH-dGuo/dGuo ratio

DNA was isolated from the brain and liver by treatment of the homogenate [10% (w/v) in 20 mM Tris-HCl buffer, pH 7.4] with 1% sodium dodecyl sulfate and 2.5 mg/ml proteinase K [36]. After extraction with phenol and then chloroform-isoamyl alcohol (24:1), DNA was precipitated by ethanol, dried, and enzymatically digested [37]. The nucleosides in a 20–50  $\mu$ l aliquot were separated by HPLC (Hewlett Packard HP 1090, Palo Alto, CA, USA) on a Nucleosil C18 reverse-phase column (250 × 4.6 mm, 5  $\mu$ m, Alltech Assoc. Inc., Deerfield, IL, USA) using 10% methanol in 50 mM potassium phosphate, pH 5.5, as the mobile phase at 1 ml/min for

Analyte	Antioxidant pretreatment				
Tissue	None	Melatonin	Vitamin C	β-Carotene	
GSH					
Brain	64 <sup>a</sup>	100 <sup>b</sup>	$76^{\rm a}$	64 <sup>a</sup>	
Lung	62 <sup>a</sup>	92 <sup>b</sup>	83	75	
Liver	81 <sup>a</sup>	100 <sup>b</sup>	90	90 <sup>a</sup>	
GSSG					
Brain	144 <sup>a</sup>	100 <sup>b</sup>	122 <sup>a</sup>	128 <sup>a</sup>	
Lung	134 <sup>a</sup>	100 <sup>b</sup>	116	122 <sup>a</sup>	
Liver	118 <sup>a</sup>	106 <sup>b</sup>	106	112	
MDA + 4-HDA					
Brain	142 <sup>a</sup>	111 <sup>b</sup>	134 <sup>a</sup>	136 <sup>a</sup>	
Lung	132 <sup>a</sup>	100 <sup>b</sup>	117	120 <sup>a</sup>	
Liver	125 <sup>a</sup>	103 <sup>b</sup>	100	99	
8-OH-dGuo/dGuo					
Brain	170 <sup>a</sup>	111 <sup>b</sup>	140 <sup>a</sup>	162 <sup>a</sup>	
Liver	139 <sup>a</sup>	113 <sup>b</sup>	110	133	

Table 3. Percentage Effects of PH<sub>3</sub> Alone and With Antioxidants on GSH, GSSG, Lipid Peroxidation Products and 8-OH-dGuo/dGuo Ratio in IP-Treated Rats (PH<sub>3</sub>/Corresponding Control ×100)

Data from Table 2.

<sup>a</sup> PH<sub>3</sub> significantly different from corresponding control without PH<sub>3</sub> (p < .05).

<sup>b</sup> Melatonin with PH<sub>3</sub> significantly different from no antioxidant with PH<sub>3</sub> (p < .05).

30 min. The DNA content of 8-OH-dGuo was analyzed by electrochemical detection (LC-4B, Bioanalytical Systems, West Lafayette, ID, USA) and of deoxyguanosine (dGuo) by UV absorbance with a diode array detector [38]. Standard mixtures of 8-OH-dGuo and dGuo were also chromatographed to generate standard curves for quantitative analysis. The amount of 8-OH-dGuo was expressed as the number for every  $10^5$  dGuo in DNA.

## Statistical analysis

Significant differences between multiple groups were determined using one-way ANOVA followed by the Bonferroni method. A probability of p < .05 was accepted as significant.

#### RESULTS

### Preliminary study

PH<sub>3</sub> adminstered ip at 2 mg/kg induced lipid peroxidation (measured as MDA plus 4-HDA) in brain within 15 min, i.e.,  $123 \pm 13\%$  (n = 6) relative to control values. The level increased further at 30 min ( $142 \pm 9\%$ ) and remained unchanged at 90 min ( $148 \pm 10\%$ ). These observations led to a more detailed study of PH<sub>3</sub>-induced oxidative damage and standardization of the conditions with sacrifice 30 min after treatment with PH<sub>3</sub>.

# Effects of $PH_3$ and antioxidants on GSH and GSSG levels

 $PH_3$  significantly decreased GSH and increased GSSG levels in all test tissues (Table 2). The antioxi-

dants in themselves did not affect the GSH and GSSG levels. Melatonin pretreatment completely abolished  $PH_3$ -induced changes in GSH and GSSG concentrations in brain and liver and ameliorated the effect in lung; in each case the reversal was statistically significant. Vitamin C reduced the  $PH_3$ -induced GSH and GSSG changes, though not to a significant degree when compared to the  $PH_3$ -treated animals without antioxidant.  $\beta$ -Carotene had little or no effect on  $PH_3$ -induced changes.

# Effects of $PH_3$ and antioxidants on MDA + 4-HDA levels

PH<sub>3</sub> significantly increased lipid peroxidation in brain, lung, and liver above the levels in control animals (Table 2). Lipid peroxidation was not affected by the antioxidants alone. The PH<sub>3</sub>-induced increase was significantly or completely blocked by melatonin pretreatment in all assayed tissues. In contrast, vitamin C and  $\beta$ -carotene did not significantly attenuate the PH<sub>3</sub>-induced increase in brain, lung, and liver.

# Effects of $PH_3$ and antioxidants on 8-OH-dGuo/dGuo ratio

PH<sub>3</sub>-induced DNA damage was evident in brain and liver by the higher 8-OH-dGuo/dGuo ratio found in treated than in control animals (Table 2). Melatonin significantly reversed the PH<sub>3</sub>-induced changes in brain and liver. Vitamin C marginally but not significantly diminished the elevated ratios.  $\beta$ -Carotene was the least effective in protecting against DNA damage.



Fig. 1. Proposed scheme for  $PH_3$ -induced oxidative damage and alleviating action of melatonin. Arrows in parentheses designate  $PH_3$ -induced increase (up) or decrease (down) in activity or amount. Asterisk with an arrow indicates that melatonin attenuates or blocks the change induced by  $PH_3$  (this study). Asterisk alone denotes that melatonin quenches ROS or stimulates an enzyme. The text provides further information on these relationships.

#### DISCUSSION

This study establishes that PH<sub>3</sub> induces oxidative damage in brain, lung, and liver of rats with partial to complete protection by melatonin and less attenuation by vitamin C and  $\beta$ -carotene (Table 3). More specifically, PH<sub>3</sub> significantly decreases GSH and increases GSSG concentrations in brain, lung and liver (36-44%, 34-38%, and 18–19%, respectively); these changes are attenuated or negated by melatonin, with much less effect of vitamin C or  $\beta$ -carotene. The findings on PH<sub>3</sub> and antioxidant effects on GSH and GSSG levels are closely paralleled by the elevation in lipid peroxidation by 42%, 32% and 25% in brain, lung, and liver, respectively, in each case largely or completely abolished by melatonin and partially by vitamin C and  $\beta$ -carotene. The 8-OHdGuo/dGuo ratios are also markedly elevated by PH<sub>3</sub> in brain and lung (70% and 39%, respectively) with significant protection by melatonin but not by vitamin C and  $\beta$ -carotene. On an overall basis, the antioxidants reduce PH<sub>3</sub>-induced oxidative damage with an effectiveness order of melatonin > vitamin C  $\geq \beta$ -carotene.

GSH is a key component of the cellular defense cascade against injury caused by ROS and the level of GSH in tissues serves as an indicator of oxidative stress. GSH is a cofactor for GSH peroxidase, which catalyzes the reduction of hydrogen peroxide to water, thereby limiting the formation of hydroxyl radical, the most toxic of the oxygen-based radicals. 8-OH-dGuo is a sensitive biomarker of ROS-mediated DNA damage [39] and is considered to be an important lesion since it is mutagenic in DNA replication [40].

The action of  $PH_3$  in rats reported here correlates well with our previous findings of increased  $H_2O_2$  production, lipid peroxidation, and oxidized DNA in Hepa 1c1c7 cells [18]. Decreased GSH in rat tissues with a concurrent rise in GSSG strongly suggests the involvement of ROS in PH<sub>3</sub> toxicity. Depletion of GSH favors lipid peroxidation and predisposes cells to oxidant damage [41]. The failure to observe PH<sub>3</sub>-induced changes in GSH levels in cultured cells [18] and insects [14,15] may relate to GSH resynthesis with the long experimental periods involved. PH<sub>3</sub> induces a higher degree of oxidative toxicity in brain and lung than in liver possibly associated with their relative oxygen consumption and unsaturated lipid content [42].

Melatonin, the most effective antioxidant under the test conditions, limits GSH depletion, GSSG formation, lipid peroxidation, and 8-OH-dGuo formation in tissues of  $PH_3$ -treated rats. These findings are consistent with previous studies where melatonin protects against oxidative damage of other ROS-generating agents, e.g. paraquat, cyanide, and kainic acid, and is a potent scavenger of hydroxyl and peroxyl radicals both in vivo and in vitro [20,23,43–45].

In conclusion, the overall findings establish that PH<sub>3</sub> induces and melatonin protects against oxidative damage

in the brain, lung and liver of rats and suggest the involvement of ROS in the genotoxicity of PH<sub>3</sub> (Fig. 1).

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## ABBREVIATIONS

AlP-aluminium phosphide

dGuo-deoxyguanosine

GSH—glutathione

GSSG-glutathione disulfide

4-HDA—4-hydroxyalkenal (such as 4-hydroxy-2(E)nonenal)

MDA—malondialdehyde

8-OH-dGuo-8-hydroxydeoxyguanosine

PH<sub>3</sub>—phosphine

- ROS-reactive oxygen species
- SOD-superoxide dismutase